

State of West Virginia Department of Administration Purchasing Division 2019 Washington Street East Post Office Box 50130 Charleston, WV 25305-0130

Request for Quotation

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RECNUMBER
LBS90130

PAGE 1

ADDRESS CORRESPONDENCE TO ATTENTION OF

ROBERTA WAGNER 804-558-0067

*906122935 262-241-7900 SEAL ANALYTICAL INC MEQUON TECHNOLOGY CENTER 10520 C BAEHR RD MEQUON WI 53092

HEALTH AND HUMAN RESOURCES ENVIRONMENTAL CHEMISTRY LAB

4710 CHIMNEY DRIVE CHARLESTON, WV 25302 304-

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10520 C BAEHR RD

53092

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RFQ COST SHEET

Bidders shall provide a cost for the following:

Automated Discrete Analyzer System

\$ 43,953,00

On-site Training (3 days at installation of equipment)

First Year Extended Warranty

3,900,00

Total Cost

\$ 47, 853.00

The award will be made to the vendor with the lowest overall total cost of the equipment which meets all requested specifications and requirements. Payment will be made in arrears.

Vendor Signature

WV-96 Rev. 10/07

AGREEMENT ADDENDUM

In the event of conflict between this addendum and the agreement, this addendum shall control:

- **DISPUTES** Any references in the agreement to arbitration or to the jurisdiction of any court are hereby deleted. Disputes arising out of the agreement shall be presented to the West Virginia Court of Claims. 1.
- HOLD HARMLESS Any clause requiring the Agency to indemnify or hold harmless any party is hereby deleted in its entirety. 2.
- GOVERNING LAW The agreement shall be governed by the laws of the State of West Virginia. This provision replaces any references to any other State's governing law. 3.
- TAXES Provisions in the agreement requiring the Agency to pay taxes are deleted. As a State entity, the Agency is exempt from Federal, State, and local taxes and will not pay taxes for any Vendor including individuals, nor will the Agency file any tax returns or reports on behalf of Vendor 4. or any other party.
- PAYMENT Any references to prepayment are deleted. Payment will be in arrears. 5.
- INTEREST Should the agreement include a provision for interest on late payments, the Agency agrees to pay the maximum legal rate under West Virginia law. All other references to interest or late charges are deleted. 6.
- RECOUPMENT Any language in the agreement waiving the Agency's right to set-off, counterclaim, recoupment, or other defense is hereby 7.
- FISCAL YEAR FUNDING Service performed under the agreement may be continued in succeeding fiscal years for the term of the agreement, contingent upon funds being appropriated by the Legislature or otherwise being available for this service. In the event funds are not appropriated or otherwise available for this service, the agreement shall terminate without penalty on June 30. After that date, the agreement becomes of no effect and is null and void. However, the Agency agrees to use its best efforts to have the amounts contemplated under the agreement included in its budget. Non-appropriation or non-funding shall not be considered an event of default. 8.
- STATUTE OF LIMITATION Any clauses limiting the time in which the Agency may bring suit against the Vendor, lessor, individual, or any 9. other party are deleted.
- SIMILAR SERVICES Any provisions limiting the Agency's right to obtain similar services or equipment in the event of default or non-funding during the term of the agreement are hereby deleted. 10.
- ATTORNEY FEES The Agency recognizes an obligation to pay attorney's fees or costs only when assessed by a court of competent jurisdiction. Any other provision is invalid and considered null and void. 11.
- ASSIGNMENT Notwithstanding any clause to the contrary, the Agency reserves the right to assign the agreement to another State of West Virginia agency, board or commission upon thirty (30) days written notice to the Vendor and Vendor shall obtain the written consent of Agency prior to assigning the agreement. 12.
- LIMITATION OF LIABILITY The Agency, as a State entity, cannot agree to assume the potential liability of a Vendor. Accordingly, any provision limiting the Vendor's liability for direct damages to a certain dollar amount or to the amount of the agreement is hereby deleted. Limitations on special, incidental or consequential damages are acceptable. In addition, any limitation is null and void to the extent that it precludes any action for injury to persons or for damages to personal property. 13.
- RIGHT TO TERMINATE Agency shall have the right to terminate the agreement upon thirty (30) days written notice to Vendor. Agency agrees to pay Vendor for services rendered or goods received prior to the effective date of termination. 14.
- TERMINATION CHARGES Any provision requiring the Agency to pay a fixed amount or liquidated damages upon termination of the agreement is hereby deleted. The Agency may only agree to reimburse a Vendor for actual costs incurred or losses sustained during the current fiscal year due to wrongful termination by the Agency prior to the end of any current agreement term. 15.
- RENEWAL Any reference to automatic renewal is hereby deleted. The agreement may be renewed only upon mutual written agreement of the 16.
- INSURANCE Any provision requiring the Agency to insure equipment or property of any kind and name the Vendor as beneficiary or as an additional insured is hereby deleted. 17.
- RIGHT TO NOTICE Any provision for repossession of equipment without notice is hereby deleted. However, the Agency does recognize a right of repossession with notice. 18.
- ACCELERATION Any reference to acceleration of payments in the event of default or non-funding is hereby deleted. 19.
- CONFIDENTIALITY: -Any provision regarding confidentiality of the terms and conditions of the agreement is hereby deleted. State contracts are public records under the West Virginia Freedom of Information Act. 20.
- AMENDMENTS All amendments, modifications, alterations or changes to the agreement shall be in writing and signed by both parties. No amendment, modification, alteration or change may be made to this addendum without the express written approval of the Purchasing Division and the Attorney General. 21.

ACCEPTED BY:

STATE OF WEST VIRGINIA	VENDOR
Spending Unit:	Company Name: <u>5 EAL ANALYTEAL</u> THE Signed: Sunal Comme
Signed:	Signed:
Title:	Title: PRESIDEAST
Date:	Date: 45/15/09

ATTACHMENT P.O.# LB390130

This agreement constitutes the entire agreement between the parties, and there are no other terms and conditions applicable to the licenses granted hereunder.

Agreed Summos/15/09	e e	
Signature Date	Signature	Date
PRESIDENT Title	Title	······································
SEAL ANALGTICAL, INC. Company Name	Agency/Divisio	 n

RFQ No. LB590130

STATE OF WEST VIRGINIA Purchasing Division

PURCHASING AFFIDAVIT

VENDOR OWING A DEBT TO THE STATE:

West Virginia Code §5A-3-10a provides that: No contract or renewal of any contract may be awarded by the state or any of its political subdivisions to any vendor or prospective vendor when the vendor or prospective vendor or a related party to the vendor or prospective vendor is a debtor and the debt owed is an amount greater than one thousand dollars in the aggregate.

PUBLIC IMPROVEMENT CONTRACTS & DRUG-FREE WORKPLACE ACT:

If this is a solicitation for a public improvement construction contract, the vendor, by its signature below, affirms that it has a written plan for a drug-free workplace policy in compliance with Article 1D, Chapter 21 of the West Virginia Code. The vendor must make said affirmation with its bid submission. Further, public improvement construction contract may not be awarded to a vendor who does not have a written plan for a drug-free workplace policy in compliance with Article 1D, Chapter 21 of the West Virginia Code and who has not submitted that plan to the appropriate contracting authority in timely fashion. For a vendor who is a subcontractor, compliance with Section 5, Article 1D, Chapter 21 of the West Virginia Code may take place before their work on the public improvement is begun.

ANTITRUST:

In submitting a bid to any agency for the state of West Virginia, the bidder offers and agrees that if the bid is accepted the bidder will convey, sell, assign or transfer to the state of West Virginia all rights, title and interest in and to all causes of action it may now or hereafter acquire under the antitrust laws of the United States and the state of West Virginia for price fixing and/or unreasonable restraints of trade relating to the particular commodities or services purchased or acquired by the state of West Virginia. Such assignment shall be made and become effective at the time the purchasing agency tenders the initial payment to the bidder.

I certify that this bid is made without prior understanding, agreement, or connection with any corporation, firm, limited liability company, partnership or person or entity submitting a bid for the same materials, supplies, equipment or services and is in all respects fair and without collusion or fraud. I further certify that I am authorized to sign the certification on behalf of the bidder or this bid.

LICENSING: APPLECATED FOR EXEMPTION FROM CERTIFICATE OF AUTHOR ITY Vendors must be licensed and in good standing in accordance with any and all state and local laws and HAS requirements by any state or local agency of West Virginia, including, but not limited to, the West Virginia $\delta \in \mathcal{EN}$ Secretary of State's Office, the West Virginia Tax Department, West Virginia Insurance Commission, or any Submitto other state agencies or political subdivision. Furthermore, the vendor must provide all necessary releases to obtain information to enable the Director or spending unit to verify that the vendor is licensed and in good standing with the above entities.

CONFIDENTIALITY:

The vendor agrees that he or she will not disclose to anyone, directly or indirectly, any such personally identifiable information or other confidential information gained from the agency, unless the individual who is the subject of the information consents to the disclosure in writing or the disclosure is made pursuant to the agency's policies, procedures and rules. Vendor further agrees to comply with the Confidentiality Policies and Information Security Accountability Requirements, set forth in http://www.state.wv.us/admin/purchase/privacy/ noticeConfidentiality.pdf.

Under penalty of law for false swearing (West Virginia Code §61-5-3), it is hereby certified that the vendor affirms and acknowledges the information in this affidavit and is in compliance with the requirements as stated.

Vendor's Name: <u>SEAL AMALYTTEAL, TWC.</u> Authorized Signature: Swall & Drum	
1 st X Q	Date: 05//5/09
Authorized Signature: Swall & Shum	Date: <u>65 //5 / 9 /</u>
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Purchasing Affidavit (Revised 01/01/09)



State of West Virginia
Department of Administration
Purchasing Division
2019 Washington Street East
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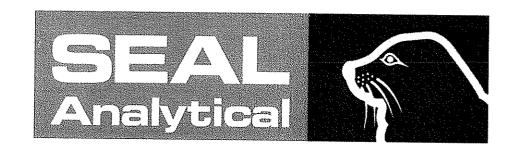
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STATE OF WEST VIRGINIA RFQ LBS90130 Automated Discrete Analyzer

RESPONSE TO BID SPECIFICATIONS

Submittals with Bid:

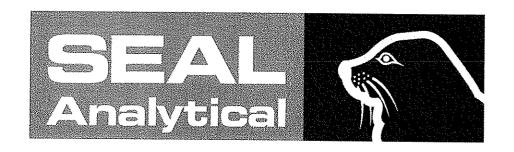
- 1. AQ2+ brochure
- 2. SEAL Analytical Who's Who and Qualifications of Technical Staff
- 3. SEAL Service and Support Contract
- 4. SEAL Analytical reference list. More references available, if requested.
- 5. SEAL Environmental Methods List
- 6. Front pages from SEAL methods per Request for Quotation
- 7. EPA letters of approval for specified methods
- 8. Two Column Comparisons for specified methods
- 9. Certificate of Insurance

The quoted AQ2+ Discrete Analyzer from SEAL Analytical meets or exceeds all the listed specifications.

Additional Notes:

INSTRUMENT SPECIFICATIONS

As the manufacturer of the original Technicon AutoAnalyzer II, we can confirm that there are certain critical parts no longer being manufactured as new.



WARRANTY REQUIREMENTS

- 1. We provide a telephone/email/fax service desk at our Milwaukee facility from 9:00 AM to 5:00 PM as part of our standard service to our customers. All service calls are logged and an in-house Engineer or Chemist will trouble-shoot the problem by telephone. Our statistics show that over 80% of user questions are diagnosed and corrected by telephone and/or email intervention, without a site visit being necessary. If this does not resolve the problem, a field based Technical Specialist is assigned to visit the site.
- 2. Technical Support Service Contracts: A full range of customer support contracts are available. (Please see attached.) It is our aim to supply a fully-installed system and provide the application, software and engineering back up to ensure the highest performance from your new analytical system.
- 3. Maintenance Service: An annual preventative maintenance service visit is provided with a purchased standard service contract or the preventative maintenance visit can be purchased separately. This preventative maintenance service would include replacement of all the standard PM parts: all the tubing in the instrument, along with all the syringe components, sampler and aspiration probes, probe flusher and lamp. As part of the PM, temperature and voltage checks are then performed. Finally, the dilutor accuracy and detector precision are checked.
- 4. On site emergency service (including travel expenses, labor and parts) is included as part of the standard warranty during the first year of use. One annual emergency visit is included as part of our premium support and service contract for year two and thereafter. For those customers with a basic support and service contract, or without a contract, the on site emergency service fee is negotiated on a case by case basis (base fees: \$400/day travel expenses plus air fare & \$150/hr labor). In some cases, it is most advantageous for the AQ2+ to be returned to our facility for repair. A loaner stock of AQ2+ analyzers, along with a parts and consumables inventory, is maintained at the Milwaukee site.
 - 5. Our <u>typical</u> response time for a Specialist to reach a site would be within 24 working hours. All support staff are our own employees, and not contract labor, and therefore would be dispatched from the Milwaukee site.

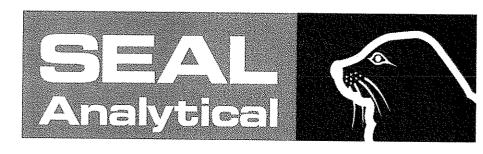
SEAL Analytical, Inc. Mequon Technology Center 10520-C Baehr Rd. Mequon, WI 53092 Phone: 262-241-7900

Fax: 262-241-7970

Email: <u>sales@seal-us.com</u> www.seal-analytical.com

Price Quotation for:

State of West Virginia
Department of Administration
Purchasing Division
2019 Washington St., East
Charleston, WV 25311



Date: 05/18/09 Valid Until: 07/18/09

Quote No: JEK-090518-RW

Phone: (304) 558-0067 Fax: (304) 558-4115

Email: rwagner@wvadmin.gov

For the Attention of: Roberta Wagner



AQ2+ Automated Multi-Chemistry Analyzer to include:

AQ2+ Chemistry Unit, 120V/60Hz (for up to 7 filter wavelengths)

Operating Software

Start-up accessories kit to include -

Lamp assembly, Peristaltic tubing, Syringe piston w/ glass body, Reaction segments, Sample cups.

Reagent containers

2x57 position sample travs

Operators', S/W & Methods, Manuals

Nitrate Cadmium Reduction Hardware \$47,500 (includes cadmium coil)

3 Days Install/Training
 12 month Warranty (parts & labor)
 Freight and Handling
 included
 \$ 800

U.S. List Price \$48,300 Less Governmental Discount – 9% \$4,347

Discounted OFFER Price \$43,953

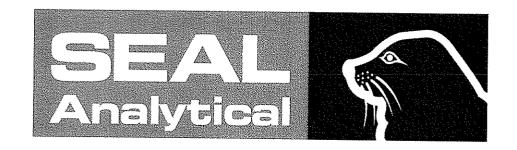
Terms: 30% payable within 7 days of placement of order

70% upon delivery, payable within 30 days

Subject to credit approval

Delivery: 4 weeks after receipt of order

Freight: FOB Destination.



WHAT IS NOT INCLUDED? - OPTIONS

Hi Spec Computer:

includes 17" flat panel monitor and laser printer

Unit PRICE

\$ 1600

Please note that the PC may be purchased by your company; however, you must meet our minimum specifications:

2.8 GHz (or above) Pentium 4 PC with: CD-RW Drive for reading software and back-up of data

512 MB RAM

40 GB Hard Disk

2 Spare USB Ports

Parallel Printer Port

Keyboard, Mouse (PS2 connection)

Flat Panel Display Monitor (capable of 128x1024 resolution at 32 bit color)

High Speed Laser Printer (6ppm)

Windows XP Home or Professional Software Package

100 Position Sample Tray

includes one bag of 2000 sample cups

Unit PRICE

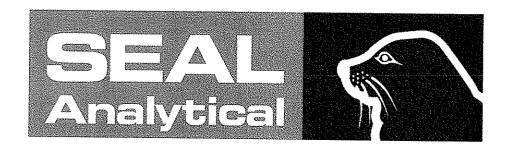
\$ 450

APC Back-UPS 900VA

Unit PRICE

\$ 320

SEAL Analytical will not assume responsibility for damage caused to instrument from power outages or power surges in the laboratory.



Summary

AQ2 Automated Multi-Chemistry Discrete Analyzer

The AQ2 Analyzer is a bench top Analyzer that has been developed to meet the specific need for a modern, automated discrete analyzer in the environmental laboratory.

This analyzer has been developed to address the needs of laboratories that require high levels of automation, a wide range of chemistries, limits of detection that ensure compliance with regulatory requirements and the advantages of integration with LIMS systems. The AQ2 is a flexible system that meets these requirements and can be configured via the industry specific software to meet the operational needs of laboratories with a wide range of analytical and throughput levels.

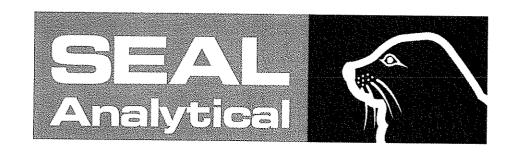


Design & Functional Specifications

The design concept of the SEAL Analyzers is to enhance productivity of analytical services in busy laboratories by streamlining the workflow of samples and information through the total analytical process.

Extensive experience in environmental laboratories has shown that the key to increasing productivity is the avoidance of multiple types of equipment and analytical procedures. In a laboratory where the routine range of colorimetric determinants requires the use of a range of instruments with restricted test menu, low throughput and multiple calibration/maintenance needs, the workflow involves several processes which are labor intensive, subject to potential error, and costly to operate. These include:

- The need to produce separate work-lists/load-lists for the different analyzers, with associated test order entry.
- The splitting of samples and maintaining integrity of identity of the sub-aliquots.
- o Separate calibration, operation, and quality control regimens on multiple analyzers.
- o Collating results form multiple analyzers for final reports, possibly with multiple interfaces to LIMS.



When these elements are rigorously costed, inclusive of personnel time, instrument maintenance costs, space requirements, duplicated consumables, I.T. costs, etc., it becomes clear that the overall cost of analysis per sample is greatly reduced by consolidating the workflow onto a single nutrients system. Thus, the design specification of the SEAL Analyzers was to develop a family of units based on a standard software platform, utilizing the same range of chemistries, which have the breadth of test menu, speed of throughput, and flexibility to process a workload that would otherwise require a combination of different contemporary analyzers and manual assay techniques.

Equally, it is a requirement that any analyzer selected meets the analytical performance criteria set out by the US EPA and other Standard bodies. Whereas, many contemporary analyzers in this market are required to operate at the extreme edge of their detection capability to meet these limits, the SEAL Analyzers routinely meet these criteria.

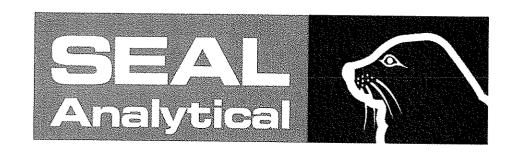
Finally, SEAL recognizes that the environment within which our systems operate is subject to change, whether from new technical demands, regulatory affairs, or changing practices of the Water & Environmental Industry. Therefore, we commit continuing resources to system development, including hardware enhancements, applications, software and chemistry methods.

Environmental Policy

SEAL is a manufacturer of high-quality automated analyzers and supplier of reagent systems for use in the Water, Soils and Environmental analysis sectors. The Company is aware that its products do have an impact on the environment in both the production and end-user stages of the product life cycle. The Company is therefore making every effort to determine the environmental impact of its operation and products and, where possible, implement a policy of reduction.

Specifically:

- Design new reagents and analyzers to take account of environmental issues such as waste production, power consumption, and low heat dissipation.
- Reformulate reagent systems to reduce harmful waste without affecting analytical performance.
- Provide full COSHH safety data on all products.
- Recycle, where practical, in-house produced waste.
- Minimize energy consumption within the factory and in the transport provided to employees.



Environmental Issues

- Discrete analysis by definition uses less than 10% of the reagents used by Continuous Flow systems.
- The Reaction Segments are manufactured from polystyrene. This material presents some issues for recycling but SEAL is currently investigating opportunities.

Warranty Statement

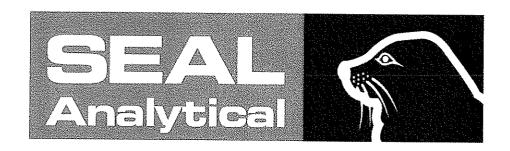
Please see below for our standard warranty statement.

AQ2 Training

The AQ2 system and software design has been strongly influenced through years of listening to SEAL users. As a result, the system is easy and intuitive to use and thus an on-site training course is conducted at installation.

The training agenda covers the following aspects of the system:

- Introduction
- Hardware overview
- Principles of operation
- Daily routine
- System software
 - Test parameters
 - Reagent parameters
 - o Standard definition
 - o AQC definition
 - o Scheduling
 - Routine running
 - Acceptance
 - Data storage routines
 - LIMS interface
- Analyzer routine maintenance
- Troubleshooting/corrective maintenance
- Method Applications



Warranty Policy:

- Subject to the conditions set out below, the Company warrants that the Goods will correspond with their specification
 at the time of delivery and will be free from defects in material and workmanship for a period of 12 months from the
 date of the initial use or 15 months from delivery, whichever is the first to expire.
- The above warranty given by the Company subject to the following conditions:
- The Company shall be under no liability in respect of any defect in the Goods arising from any drawings, design or specification supplied by the Buyer;
- The Company shall be under no liability in respect of any defect arising from fair wear and tear, willful damage, negligence, abnormal working conditions, failure to follow the Company's instruction (whether oral or in writing), misuse or alteration or repair of the Goods without the Company's approval;
- The Company shall be under no liability under the above warranty (or any other warranty, conditions of guarantee) if the total price of the Goods has not been paid by the due date of payment;
- The above warranty does not extend to parts, materials or equipment not manufactured by the Company, in respect of which the Buyer shall only be entitled to the benefit of any such warranty or guarantee as is given by the manufacturer to the Company.
- Any claim by the Buyer which is based on any defect in the quality or condition of the Goods or their failure to correspond with specification shall (whether or not delivery is refused by the Buyer) be notified by the Company within 7 days from the date of delivery (where the defect or failure was not apparent on reasonable inspection) within a reasonable time within discovery of the defect or failure. If delivery is not refused and the Buyer does not notify the Company accordingly, the Buyer shall not be entitled to reject the Goods and the Company shall have no liability for such defects or failure and the Buyer shall be bound to pay the price as if the Goods had been delivered in accordance with the Contract.
- Where any valid claim in respect of any of the Goods which is based on any defect in the quality of condition of the Goods or their failure to meet specification is notified to the Company in accordance with these Conditions, the Company shall be entitled to replace the Goods (or the part in question) free of charge, at the Company's sole discretion, refund to the Buyer the price of the Goods (or a proportionate part of the price), but the Company shall have no further liability to the Buyer.
- Except in respect of the death or personal injury caused by the Company's negligence, the Company shall not be liable
 to the Buyer by reason of any representation, or any implied warranty, condition or other term, or any duty at common
 law, or under the express terms of the Contract, for any consequential loss or damage (whether for loss of profit or
 otherwise), costs expenses or other claims for consequential compensation whatsoever (and whether caused by the
 negligence of the Company, its employees or agents or otherwise) which arise out of or in connection with the supply
 of Goods or their use resale by the Buyer except as expressly provided in these Terms.
- The Company shall not be liable to the Buyer or be deemed to be in breech of the Contract by reason of any delay in performing, or any failure to perform, any of the Company's obligations in relation to the Goods, if the delay or failure was due to any cause beyond the Company's reasonable control. Without prejudice to the generality of the foregoing, the following shall be regarded as causes beyond the Company's reasonable control; Act of God, explosion, flood, tempest, fire or accident; war or threat of war, sabotage, insurrection, civil disturbance or requisition, acts, requisition, regulations, bye laws, prohibition or measures of any kind of the part of any Governmental, Local Authority; import or export regulations or embargo's, strike, lock-outs or other industrial actions or trade disputes (whether involving employees of the Company or of a third party); difficulties in obtaining raw materials, labor, fuel, parts or machinery; power failure or breakdown in machinery.



ENVIRONMENTAL METHODS LIST — USEPA Rev 11 Method Detection Limits are calculated using USEPA procedure 40 CFR, Part 136, Appendix B



ANALYTE	METHOD DESCRIPTION	METHOD NUMBER	METHOD DETECTION LIMIT	EOUIVAI ENCE
		EPA-100-A Rev. 2	3.5 mg CaCO ₃ /L (Range: 5 to 80 mg/L)	
ALKALINITY	Buffered methyl orange color reduction	EPA-101-A Rev. 2	8 mg CaCO ₃ /L (Range: 15 to 200 mg/L)	EPA 310.2
	Table 1 and	EPA-102-A Rev. 2	27 mg CaCO ₃ /L (Range: 44 to 550 mg/L)	
AMMONIA	Alkaline phenate method with hypochlorite and sodium nitroprusside	EPA-103-A Rev. 6	0.007 mg N/L (Range: 0.02 to 2.0 mg N/L)	EPA 350.1
- Provider	(indophenol blue)	EPA-129-A Rev. 5	0.05 mg N/L (Range: 0.2 to 10 mg N/L)	Std. Methods 4500-NH ₃ G (19 th ,20 th)
AMMONIA	Alkaline phenate method with hypochlorite and sodium nitroprusside (indophenol blue). This is a brackish method.	EPA-104-A Rev. 3	0.02 mg N/L (Range: 0.05 to 2.0 mg N/L)	EPA 350.1 Std. Methods 4500-NH ₃ G (19th,20 th)
CHLORIDE	Mercuric thiocyanate reaction in the	EPA-105-A Rev. 4	0.3 mg/L (Range: 2 to 100 mg/L)	Std Methods 4500 O'E
	presence of terric nitrate	EPA-124-A Rev. 4	0.4 mg/L (Range: 5 to 200 mg/L)	(18 th , 19 th , 20 th)
CHROMIUM, Hexavalent	Colorimetric	EPA-108-A Rev. 1	0.001 mg/L (Range: 0.025 to 0.5 mg Cr(VI)/L)	EDA 218 4
		EPA-109-A Rev. 1	0.011 mg/L (Range: 0.3 to 5.0 mg Cr(VI)/L)	Std, Methods 3500-Cr B (20th)
COLOR	Platinum-cobalt standard comparison	EPA-140-A Rev. 1	2 Color Units (Range: 5 to 150 Color Units)	Std. Methods 2120 B (18th, 19th, 20th)
CYANIDE	Amenable to chlorination (Manual distillation required)	EPA-107-A Rev. 5	0.0004 mg CN/L (Range: 0.002 to 0.3 mg CN/L)	EPA 335.1 Std. Methods 4500-CN G (18th, 19th, 20th)
CYANIDE	Chloramine-T with pyridine barbituric acid color reaction (Manual distillation required)	EPA-107-A Rev. 5	0.0004 mg CN/L (Range: 0.002 to 0.3 mg CN/L)	EPA 335.2 Std. Methods 4500-CN E (18 th ,19 th ,20 th)
CYANIDE	Chloramine-T with pyridine barbituric acid color reaction (Manual distillation required)	EPA-130-A Rev. 3	0.0005 mg CN/L (Range: 0.003 to 0.25 mg CN/L)	EPA 335.4 Std. Methods 4500-CN E (18th, 19th, 20th)
CYANIDE	Amenable to chlorination (Without distillation)	EPA-133-A Rev. 1	0.0004 mg CN/L (Range: 0.002 to 0.3 mg CN/L)	Std. Methods 4500-CN H (20 th)

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ENVIRONMENTAL METHODS LIST — USEPA Rev 11 Method Detection Limits are calculated using USEPA procedure 40 CFR, Part 136, Appendix B

ANALYTE METHOD DESCRIPTION METHOD DUBECRIPTION METHOD DEJCO3/L (Range 25) HARDNESS, Total Calmagite indicator reaction EPA-105-A Rev. 5 1 mg CaCO3/L (Range 25) NITROGEN, With Riskland (TKN) Wighdaid digests (Fig catalyst) are reacted with a Relative salicylate in the presence of Total Keldeni (TKN) EPA-110-A Rev. 5 0.03 mg NL (Range: 0.1 to Digestion required) NITROGEN, With a Richard (TKN) Mighaid digests (Fig catalyst) are reacted with a Richard (TKN) EPA-110-A Rev. 5 0.03 mg NL (Range: 0.1 to Digestion required) NITRATE + NITRITE Cadmium coil reduction followed by Intratte + NITRITE Cadmium coil reduction followed by EPA-112-A Rev. 5 0.003 mg NL (Range: 0.012 to Digestion required) NITRATE + NITRITE NITRATE Sulfanilarinde reaction in the presence of dihydrochloride enaction in the presence of differ couplydateleantinony with ascorbic enaction in the presence of		THE PARTY OF THE P			
Calmaglie indicator reaction Kjeldahi digests (Hg catalyst) are reacted with alkaline salicylate in the presence of hypochlorite and sodium nitroprusside (Digestion required) Kjeldahi digests (Cu catalyst) are reacted with alkaline salicylate in the presence of hypochlorite and sodium nitroprusside (Digestion required) Kjeldahi digests (Cu catalyst) are reacted hypochlorite and sodium nitroprusside (Digestion required) Kjeldahi digests (Cu catalyst) are reacted hypochlorite and sodium nitroprusside (Digestion required) Kjeldahi digests (Cu catalyst) are reacted hypochlorite and sodium nitroprusside (Digestion required) Cadmium coil reduction followed by EPA-127-A Rev. 5 Cadmium coil reduction followed by EPA-126-A Rev. 5 Cadmium coil reduction in the presence of EPA-132-A Rev. 1 EPA-137-A Rev. 3 Cadmium de reaction in the presence of EPA-137-A Rev. 1 Calminamide reaction in the presence of dihydrochloride Sulfanilamide reaction in the presence of dihydrochloride Sample distillation required) Acidic molybdate/antimony with ascorbic EPA-118-A Rev. 5 EPA-118-A Rev. 5 Acidic molybdate/antimony with ascorbic EPA-118-A Rev. 5 EPA-118-A Rev. 5 EPA-118-A Rev. 5 EPA-118-A Rev. 5	ANALYTE	METHOD DESCRIPTION	METHOD NUMBER	METHOD DETECTION LIMIT	FOLIVAL ENCE
Kjeldahl digests (Hg catalyst) are reacted with alkaline salicylate in the presence of hypochlorite and sodium nitroprusside (Digestion required) Kjeldahl digests (Cu catalyst) are reacted with alkaline salicylate in the presence of hypochlorite and sodium nitroprusside (Digestion required) Kjeldahl digests (Cu catalyst) are reacted with alkaline salicylate in the presence of hypochlorite and sodium nitroprusside (Digestion required) Cadmium coil reduction followed by Sulfanilamide reaction in the presence of Gadmium coil reduction followed by Sulfanilamide reaction in the presence of Gadmium coil reduction in the presence of M-(1-naphthylethylenediamine) Sulfanilamide reaction in the presence of Gilydrochloride (midazole buffer used) Sulfanilamide reaction in the presence of Gilydrochloride (midazole buffer used) Sulfanilamide reaction in the presence of Gilydrochloride (midazole buffer used) Sulfanilamide reaction in the presence of Gilydrochloride (midazole buffer used) Sulfanilamide reaction in the presence of Gilydrochloride (midazole buffer used) Sulfanilamide reaction in the presence of Gilydrochloride (midazole buffer used) Sulfanilamide reaction in the presence of Gilydrochloride (midazole buffer used) Santhe distillation required) Acidic molybdate/antimony with ascorbic EPA-113-A Rev. 5 Acidic molybdate/antimony with ascorbic acid reduction (phosphomolybdate) Repa-113-A Rev. 5 Acidic molybdate/antimony with ascorbic EPA-113-A Rev. 5 Acidic molybdate/antimony with ascorbic acid reduction (Manual persulfate digestion required) EPA-113-A Rev. 5 Acidic molybdate/antimony with ascorbic EPA-113-A Rev. 5	HARDNESS, Total	Calmagite indicator reaction	EPA-106-A Rev. 2	11 mg CaCO ₃ /L. (Range: 25 to 400 mg/L.)	EPA 130.1
Nipochlorite and sodium nitroprusside (Digestion required) Kieldahl digests (Cu catalyst) are reacted with alkaline ferricyanide and sodium nitroprusside (Digestion required) Cadmium coil reduction followed by sulfanilamide reaction in the presence of dihydrochloride (Imidazole buffer used) Sulfanilamide reaction in the presence of dihydrochloride (Imidazole buffer used) Sulfanilamide reaction in the presence of dihydrochloride Sample distillates are reacted with alkaline ferricyanide and 4-aminoantipyrine (Manual distillation required) Acidic molybdate/antimony with ascorbic acid reduction Manual persulfate digestion required) Acidic molybdate/antimony with ascorbic acid reduction (Manual persulfate digestion required) EPA-113-A Rev. 3 EPA-114-A Rev. 3 EPA-115-A Rev. 3 GEPA-115-A Rev. 5 Acidic molybdate/antimony with ascorbic acid reduction GEPA-116-A Rev. 5 Acidic molybdate/antimony with ascorbic acid reduction GEPA-119-A Rev. 5	NITROGEN,	Kjeldahl digests (Hg catalyst) are reacted with alkaline salicylate in the presence of	EPA-125-A Rev. 5	0.03 mg N/L (Range: 0.1 to 4.0 mg N/L)	and the same of th
Kjeldahl digests (Cu catalyst) are reacted with alkaline salicylate in the presence of hypochlorite and sodium nitroprusside (Digestion required) Cadmium coil reduction followed by suffamilamide reaction in the presence of alitydrochloride (Inidazcle buffer used) Sulfamilamide reaction in the presence of alitydrochloride (Inidazcle buffer used) Sulfamilamide reaction in the presence of dihydrochloride (Inidazcle buffer used) Sulfamilamide reaction in the presence of dihydrochloride (Inidazcle buffer used) Sulfamilamide reaction in the presence of dihydrochloride Sample distillates are reacted with alkaline ferricyanide and 4-aminoantipyrine acid reduction (phosphomolybdenum blue) Acidic molybdate/antimony with ascorbic acid reduction (Manual persulfate digestion required) EPA-113-A Rev. 3 EPA-117-A Rev. 4 EPA-117-A Rev. 4 EPA-117-A Rev. 3 EPA-117-A Rev. 5 Acidic molybdate/antimony with ascorbic EPA-118-A Rev. 5 Acidic molybdate/antimony with ascorbic acid reduction (Manual persulfate digestion required) EPA-119-A Rev. 5	Total Kjeldahi (TKN)	hypochlorite and sodium nitroprusside (Digestion required)	EPA-110-A Rev. 5	0.2 mg N/L (Range: 0.5 to 24 mg N/L)	EPA 351.2, version 2 (1993)
Cadmium coil reduction followed by suffamilamide reaction in the presence of dihydrochloride (Imidazole buffer used) Sulfamilamide reaction in the presence of dihydrochloride (Imidazole buffer used) Sulfamilamide reaction in the presence of dihydrochloride (Imidazole buffer used) Sulfamilamide reaction in the presence of dihydrochloride (Imidazole buffer used) Sulfamilamide reaction in the presence of dihydrochloride Sulfamilamide reaction in the presence of dihydrochloride Sample distillates are reacted with alkaline ferricyanide and 4-aminoantipyrine Acidic molybdate/antimony with ascorbic acid reduction (phosphomolybdenum blue) Acidic molybdate/antimony with ascorbic acid reduction (Manual persulfate digestion required) Acidic molybdate/antimony with ascorbic acid reduction (Manual persulfate digestion required) EPA-113-A Rev. 3 EPA-117-A Rev. 4 EPA-116-A Rev. 3 EPA-117-A Rev. 4 EPA-118-A Rev. 3 EPA-118-A Rev. 3 EPA-118-A Rev. 5 Acidic molybdate/antimony with ascorbic acid reduction EPA-118-A Rev. 5 Acidic molybdate/antimony with ascorbic acid reduction EPA-119-A Rev. 5	NITROGEN, Total Kieldan (TKN)	Kjeldahl digests (Cu catalyst) are reacted with alkaline salicylate in the presence of	EPA-111-A Rev. 4	0.035 mg N/L (Range: 0.1 to 4.0 mg N/L)	and the state of t
Cadmium coil reduction followed by sulfanilamide reaction in the presence of dihydrochloride cadmium coil reduction followed by sulfanilamide reaction in the presence of cadmium coil reduction followed by sulfanilamide reaction in the presence of dihydrochloride (Imidazole buffer used) Sulfanilamide reaction in the presence of dihydrochloride (Imidazole buffer used) Sulfanilamide reaction in the presence of dihydrochloride Sample distillates are reacted with alkaline ferricyanide and 4-aminoantipyrine (Imidazole with acidic molybdate/antimony with ascorbic acid reduction (phosphomolybdenum blue) EPA-113-A Rev. 3 EPA-115-A Rev. 3 EPA-117-A Rev. 4 EPA-113-A Rev. 3 EPA-117-A Rev. 5 Acidic molybdate/antimony with ascorbic acid reduction (phosphomolybdenum blue) EPA-118-A Rev. 5 Acidic molybdate/antimony with ascorbic EPA-119-A Rev. 5 Acidic molybdate/antimony with ascorbic acid reduction (phosphomolybdenum blue)		rypochione and sodium nitroprusside (Digestion required)	EPA-136-A Rev. 3	0.15 mg N/L (Range: 0.4 to 24 mg N/L)	EPA 351.2, version 2 (1993)
N-(1-naphthylethylethylethylethylethylethylethyl	NITRATE + NITRITE	Cadmium coil reduction followed by sulfaniamide reaction in the presence of	EPA-127-A Rev. 5	0.003 mg N/L (Range: 0.012 to 2.0 mg N/L)	TO A COLUMN
Cadmium coil reduction followed by sulfamilamide reaction in the presence of dihydrochloride (Imidazole buffer used) Sulfamilamide reaction in the presence of dihydrochloride Sample distillates are reacted with alkaline ferricyanide and 4-aminoantipyrine (Manual distillation required) Acidic molybdate/antimony with ascorbic acid reduction (phosphomolybdenum blue) Acidic molybdate/antimony with ascorbic acid reduction (Manual persulfate digestion required) EPA-113-A Rev. 3 EPA-115-A Rev. 3 EPA-117-A Rev. 4 EPA-118-A Rev. 3 EPA-118-A Rev. 3 EPA-118-A Rev. 5 Acidic molybdate/antimony with ascorbic acid reduction (Manual persulfate digestion required) EPA-119-A Rev. 5 Acidic molybdate/antimony with ascorbic acid reduction (Manual persulfate digestion required)		N-(1-naphthylethylenediamine)	EPA-126-A Rev. 5	0.006 mg N/L (Range: 0.03 to 4.5 mg N/L)	EPA 353.2 Std. Methods 4500-NO ₃ F
Cadmium coil reduction followed by suffanilamide reaction in the presence of alitydrochloride (Imidazole buffer used) N-(1-naphthylethylenediamine) Acidic molybdate/antitmony with ascorbic acid reduction Cadmium coil reduction for the presence of alitydrochloride Sample distillation required) Acidic molybdate/antitmony with ascorbic acid reduction Cadmium coil reduction followed by suffanilamide reaction in the presence of alitydrochloride Sample distillates are reacted with alkaline ferricyanide and 4-aminoantipyrine (Manual distillation required) Acidic molybdate/antitmony with ascorbic acid reduction (Manual persulfate digestion required) EPA-118-A Rev. 5 Acidic molybdate/antitmony with ascorbic acid reduction (Manual persulfate digestion required) EPA-119-A Rev. 5 Acidic molybdate/antitmony with ascorbic acid reduction (Manual persulfate digestion required)	**************************************	ani no include	EPA-114-A Rev. 6	0.03 mg N/L (Range: 0.25 to 15 mg N/L)	(18",19",20")
Sulfanilamide reaction in the presence of dihydrochloride Sulfanilamide reaction in the presence of dihydrochloride Sulfanilamide reaction in the presence of dihydrochloride Sample distillates are reacted with alkaline ferricyanide and 4-aminoantipyrine (Manual distillation required) Acidic molybdate/antimony with ascorbic acid reduction (phosphomolybdenum blue) Acidic molybdate/antimony with ascorbic acid reduction (Manual persulfate digestion required) EPA-113-A Rev. 3 EPA-113-A Rev. 3 EPA-113-A Rev. 3 EPA-113-A Rev. 5 Acidic molybdate/antimony with ascorbic acid reduction (Manual persulfate digestion required) EPA-113-A Rev. 5 Acidic molybdate/antimony with ascorbic acid reduction EPA-113-A Rev. 5 Acidic molybdate/antimony with ascorbic acid reduction EPA-113-A Rev. 5	NITRATE + NITRITE	Cadmium coil reduction followed by sulfanilamide reaction in the presence of N-(1-naphthylethylenediamine) dihydrochloride (Imidazole buffer used)	EPA-132-A Rev. 1	0.004 mg N/L (Range: 0.012 to 2.0 mg N/L)	N/A
Sulfanilamide reaction in the presence of harmony with ascorbic acid reduction (Manual presultation required) Acidic molybdate/antimony with ascorbic acid reduction (Manual persultation required) Acidic molybdate/antimony with ascorbic acid reduction (Manual persulfate digestion required) Acidic molybdate/antimony with ascorbic acid reduction (Manual persulfate digestion required) EPA-113-A Rev. 3 EPA-113-A Rev. 5 Acidic molybdate/antimony with ascorbic acid reduction (Manual persulfate digestion required) EPA-113-A Rev. 5 Acidic molybdate/antimony with ascorbic acid reduction (Manual persulfate digestion required)	NITRITE	Sulfanilamide reaction in the presence of N-(1-naphthylethylenediamine)	EPA-115-A Rev. 3	0.0025 mg N/L (Range: 0.015 to 1.2 mg N/L)	EPA 353.2
Sulfanilamide reaction in the presence of N-(1-naphthylethylenediamine) Allydrochloride Sample distillates are reacted with alkaline ferricyanide and 4-aminoantipyrine (Manual distillation required) Acidic molybdate/antimony with ascorbic acid reduction (phosphomolybdenum blue) Acidic molybdate/antimony with ascorbic acid reduction (Manual persulfate digestion required) EPA-118-A Rev. 3 Acidic molybdate/antimony with ascorbic acid reduction (Manual persulfate digestion required) EPA-119-A Rev. 5		dihydrochloride	EPA-137-A Rev. 1	0.0006 mg N/L (Range: 0.002 to 0.2 mg N/L)	Std. Methods 4500-NO ₃ F
Sample distillates are reacted with alkaline ferricyanide and 4-aminoantipyrine (Manual distillation required) Acidic molybdate/antimony with ascorbic blue) Acidic molybdate/antimony with ascorbic acid reduction (Manual persulfate digestion required) Acidic molybdate/antimony with ascorbic acid reduction (Manual persulfate digestion required) EPA-118-A Rev. 3 EPA-118-A Rev. 5 EPA-119-A Rev. 5	NITRITE	Suffanilamide reaction in the presence of N-(1-naphthylethylenediamine) dihydrochloride	EPA-116-A Rev 3	0.0005 mg N/L (Range: 0.002 to 0.24 mg N/L)	EPA 354.1 Std. Methods 4500-NO ₂ B (18 th 19 th 20 th)
Acidic molybdate/antimony with ascorbic acid reduction (phosphomolybdenum blue) Acidic molybdate/antimony with ascorbic acid reduction (Manual persulfate digestion required) Acidic molybdate/antimony with ascorbic acid reduction (Manual persulfate digestion required)	PHENOLICS	Sample distillates are reacted with alkaline ferricyanide and 4-aminoantipyrine (Manual distillation required)	EPA-117-A Rev. 4	0.003 mg/L (Range: 0.01 to 0.4 mg phenol/L)	EPA 420.4
Acidic molybdate/antimony with ascorbic EPA-119-A Rev. 5 acid reduction (Manual persulfate digestion required) EPA-134-A Rev. 2	PHOSPHATE, Ortho	Acidic molybdate/antimony with ascorbic acid reduction (phosphomolybdenum	EPA-118-A Rev. 3	0.0015 mg P/L (Range: 0.005 to 1.0 mg P/L)	EPA 365,1
Acidic molybdate/antimony with ascorbic EPA-119-A Rev. 5 acid reduction (Manual persulfate digestion required) EPA-134-A Rev. 2		(enld	EPA-128-A Rev. 5	0.005 mg P/L (Range: 0.32 to 20 mg P/L)	Std. Methods 4500-P F (18th, 19th, 20th)
EPA-134-A Rev. 2	PHOSPHORUS, Total (TP)	Acidic molybdate/antimony with ascorbic acid reduction	EPA-119-A Rev. 5	0.002 mg P/L (Range: 0.01 to 1.0 mg P/L)	EPA 365.1
Addition the state of the state		(Manual persulfate digestion required)	EPA-134-A Rev. 2	0.007 mg P/L (Range: 0.063 to 5.0 mg P/L)	Std. Methods 4500-P B, F (18th, 19th, 20th)

ENVIRONMENTAL METHODS LIST — USEPA Rev 11 KAN Method Detection Limits are calculated using USEPA procedure 40 CFR, Part 136, Appendix B

The second secon				
ANALYTE	METHOD DESCRIPTION	METHOD NUMBER	METHOD DETECTION LIMIT	FOILVA! ENDE
PHOSPHORUS, Total Kjeldahl (TKP)	Kjeldahl digests (Hg catalyst) are reacted with acidic molybdate/antimony with ascorbic acid reduction	EPA-120-A Rev. 4	0.007 mg P/L (Range: 0.04 to 3.2 mg P/L)	EPA 365.4
PHOSPHORUS, Total Kieldahi /TKB)	Kjeldahl digests (Cu catalyst) are reacted with acidic molybdate/antimony with	EPA-135-A Rev. 1	0.009 mg P/L (Range: 0.04 to 3.2 mg P/L)	To a state of the
i viai iyotaalii (INF)	ascorbic acid reduction. Method range depends on digestion protocol	EPA-138-A	TBD mg P/L (Range: 0.05 to 3.0 mg P/L)	N/A
SILICA (Reactive silica)	Acidic molybdate, no reduction (molybdo-silicic acid)	EPA-121-A Rev. 1	0.1 mg/L (Range: 0.25 to 25 mg/L)	Std. Methods 4500-SiO ₂ C
SILICA (Reactive silica)	Acidic molybdate with ANSA reduction (silico-molybdenum blue)	EPA-122-A Rev. 5	0.025 mg/L (Range: 0.1 to 10 mg/L)	(20) Std. Methods 4500-SiO ₂ D (20 th)
SULFATE	Barium chloride turbidimetric method	EPA-123-A Rev. 4	1.0 mg/L (Range: 5 to 40 mg/L)	ASTM D516-90, 02
				•

SEAL Analytical is continually developing methods. Please note that others may exist.

If you do not see your chosen method on this list, please consult your SEAL Analytical Technical Support Team at:

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NITRATE-N + NITRITE-N IN DRINKING AND SURFACE WATERS, AND DOMESTIC AND INDUSTRIAL WASTES

AQ2 METHOD NO: EPA-114-A Rev. 6

TYPE OF SAMPLE

This method is applicable to sewage, effluents, raw and drinking waters. This method is approved for the Clean Water Act for use in wastewater compliance monitoring under the National Pollutant Discharge Elimination System (NPDES) and for use in NPDWR (National Primary Drinking Water Regulations) compliance monitoring. The laboratory should consult the most recent Code of Federal Regulations for applicability and approval by the Clean Water and Safe Drinking Water Acts (40 CFR parts 136 and 141, respectively).

SUBSTANCE DETERMINED

Nitrate and nitrite ions

RANGE OF APPLICATION

0.25 - 15 mg N/L

LIMIT OF DETECTION BY USEPA PROCEDURE

MDL = 0.03 mg N/L

METHOD PRINCIPLE - USEPA METHOD 353.2

Nitrate is reduced by copperized cadmium to nitrite, which reacts with sulfanilamide to form a diazonium compound which, in dilute phosphoric acid, couples with N-(1-naphthyl)-ethylenediamine dihydrochloride to form a reddish-purple azo dye. This is measured spectrophotometrically at 520 nm. Separate rather than combined nitrate/nitrite values are obtained by running the samples on sequential tests, i.e., first with, and then without the copperized cadmium reduction step.

INTERFERENCES

Since nitrate-nitrogen is dissolved, the sample may be pre-filtered. Low results may be obtained for samples that contain high concentrations of iron, copper or other metals. EDTA is added to the reagents to eliminate this interference. Samples that contain oil and grease may coat the copperized surface of the cadmium. This is eliminated by pre-extracting the sample with an organic solvent. If fouled, the cadmium is regenerated using an automated utility in AQ2 software. Residual chlorine can produce negative interference by decreasing reduction efficiency. Before analysis, samples should be checked and if required, dechlorinated with sodium thiosulfate.

REFERENCES

Oxidized Nitrogen in Waters 1981, HMSO. (Series Methods for the Examination of Waters and Associated Materials)

Methods for the Determination of Inorganic Substances in Environmental Samples, USEPA 600/R-93/100, August 1993: Method 353.2, Revision 2.0

SEAL Applications Revision Date: March 3, 2006

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EPA-114-A Rev 6

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UNITED STATES ENVIRONMENTAL PROTECTION AGENCY WASHINGTON, D.C. 20460

OFFICE OF WATER

MAY 19 2005

Jessica Ammerman Technical Support Director SEAL Analytical, Inc. 1492 W. Mequon Road Mequon, WI 53092

Dear Ms. Ammerman:

The Statistics and Analytical Support Branch (SASB) and the Office of Ground Water and Drinking Water's Technical Support Center (OGWDW/TSC) have completed their technical review of the materials submitted to support approval of SEAL Method EPA-114-A [Revision Date, March 28, 2005] (ATP Case Nos. D04-0014 and N04-0014). This method was submitted as an alternate test procedure (ATP) to determine combined nitrate/nitrite or nitrate and nitrite singly for use in wastewater compliance monitoring under National Pollutant Discharge Elimination System (NPDES) regulations and in drinking water compliance monitoring under national primary drinking water regulations (NPDWR).

We are pleased to inform you that, in the judgement of our technical staff, SEAL Method EPA-114-A is an acceptable version of approved revisions of EPA Method 353.2. EPA Method 353.2 is listed at Title 40 of the Code of Federal Regulations (40 CFR) Part 136 for NPDES compliance monitoring¹ and 40 CFR Part 141 for NPDWR compliance monitoring. Seal Method EPA-114-A determines combined nitrate/nitrite or nitrate and nitrite singly by colorimetry using discreet analysis. A filtered sample is passed through a copperized-cadmium column to reduce nitrate to nitrite. The nitrite originally present in the sample plus reduced nitrate reacts with sulfanilamide under acidic conditions to form a diazonium compound. This compound is then coupled with N-(1-naphthyl) ethylenediamine dihydrochloride to form a reddish-purple azo dye that is measured using a colorimeter. Separate, rather than combined nitrate/nitrite results, are obtained by carrying out the procedure first with, and then without, the copper-cadmium reduction step (to measure nitrate/nitrite and nitrite respectively)and subtracting the result obtained for nitrite from the result obtained for combined nitrate/nitrite to obtain the result for nitrate. The procedures used are equivalent to those used in EPA-approved colorimetric methods

¹EPA Method 353.2 is not currently listed at 40 CFR 136 for determination of nitrate and nitrite singly. However, EPA has proposed adding EPA Method 353.2 to 40 CFR 136 for these analytes, and previously has recommended the use of EPA Method 353.2 for determination of nitrate and nitrite singly (by analyzing NO₃ + NO₂, and subtracting NO₂) for this purpose. EPA's past recommendations extend to SEAL Method EPA-114-A.

such as EPA Method 353.2. Accordingly, SEAL Method EPA-114-A [Revision Date, March 28, 2005] may be used in place of other NPDES and NPDWR-approved colorimetric methods for determination of combined nitrate/nitrite or nitrate and nitrite singly.

We greatly appreciate SEAL's interest in the development of environmental monitoring methods. If you have any questions regarding this ATP application, please contact William Telliard of SASB at telliard.william@epa.gov or at 202/566-1061 for wastewater or Herb Brass of OGWDW/TSC at brass.herb@epa.gov or at 513/569-7936 for drinking water at your convenience.

Sincerely,

William A. Telliard

Director of Analytical Methods

Engineering and Analysis Division (4303T)

Herb Brass, Ph.D.

Technical Support Center (MS-140)

Office of Ground Water and Drinking Water

cc:

USEPA Regional Administrators (all Regions)

Quality Assurance Managers (all Regions)

ATP Coordinators (all Regions)

Water Management Division Directors (all Regions)

Gregory J. Carroll, USEPA, OGWDW

Lillian Holmes, USEPA, OGWDW/TSC

Robin K. Oshiro, USEPA, EAD

Danielle Carter, CSC, SCC





Web: www.seal-analytical.com

NITRATE-N + NITRITE-N IN DRINKING AND SURFACE WATERS, AND DOMESTIC AND INDUSTRIAL WASTES

AQ2 METHOD NO: EPA-126-A Rev. 5

TYPE OF SAMPLE

This method is applicable to sewage, effluents, raw and drinking waters. This method is approved for the Clean Water Act for use in wastewater compliance monitoring under the National Pollutant Discharge Elimination System (NPDES) and for use in NPDWR (National Primary Drinking Water Regulations) compliance monitoring. The laboratory should consult the most recent Code of Federal Regulations for applicability and approval by the Clean Water and Safe Drinking Water Acts (40 CFR parts 136 and 141, respectively).

SUBSTANCE DETERMINED

Nitrate and nitrite ions

RANGE OF APPLICATION

0.03 - 4.5 mg N/L

LIMIT OF DETECTION BY USEPA PROCEDURE

MDL = 0.006 mg N/L

METHOD PRINCIPLE - USEPA METHOD 353.2

Nitrate is reduced by copperized cadmium to nitrite, which reacts with sulfanilamide to form a diazonium compound which, in dilute phosphoric acid, couples with N-(1-naphthyl)-ethylenediamine dihydrochloride to form a reddish-purple azo dye. This is measured spectrophotometrically at 520 nm. Separate rather than combined nitrate/nitrite values are obtained by running the samples on sequential tests, i.e., first with, and then without the copperized cadmium reduction step.

INTERFERENCES

Since nitrate-nitrogen is dissolved, the sample may be pre-filtered. Low results may be obtained for samples that contain high concentrations of iron, copper or other metals. EDTA is added to the reagents to eliminate this interference. Samples that contain oil and grease may coat the copperized surface of the cadmium. This is eliminated by pre-extracting the sample with an organic solvent. If fouled, the cadmium is regenerated using an automated utility in AQ2 software. Residual chlorine can produce negative interference by decreasing reduction efficiency. Before analysis, samples should be checked and if required, dechlorinated with sodium thiosulfate.

REFERENCES

Oxidized Nitrogen in Waters 1981, HMSO. (Series Methods for the Examination of Waters and Associated Materials)

Methods for the Determination of Inorganic Substances in Environmental Samples, USEPA 600/R-93/100, 1993: Method 353.2, Revision 2.0.

SEAL Applications Revision Date: March 3, 2006

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EPA-126-A Rev 5

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UNITED STATES ENVIRONMENTAL PROTECTION AGENCY

WASHINGTON, D.C. 20460

JUN - 7 2004

OFFICE OF WATER

Laura Kohl
Technical Support Director
SEAL Analytical, Inc.
1492 W. Mequon Road
Mequon, WI 53092

Dear Ms. Kohl:

The Statistics and Analytical Support Branch (SASB) and the Office of Ground Water and Drinking Water's Technical Support Center (OGWDW/TSC) have completed their technical review of the materials submitted in support of the following applications for approval of alternate test procedures (ATPs) for use in compliance monitoring under National Pollutant Discharge Elimination System (NPDES) Regulations and National Primary Drinking Water Regulations (NPDWR):

Nethod Number Revision Date	Analytes	ATP Case No.s	
2-013-1-L [October 2003]	nitrite (low range)	D02-0006, N02-0006	
2-013-1-H [July 2003]	nitrite (high range)	D02-0007, N02-0007	

We are pleased to inform you that in the judgement of our technical staff the methods listed above are acceptable versions of EPA-approved methods listed at Title 40 of the Code of Federal Regulations (CFR) Parts 136 and 141 for determining nitrite in wastewater and drinking water, respectively. Both of the methods listed above determine nitrite by colorimetry using discreet analysis. Nitrite in a sample reacts with sulfanilamide under acidic conditions to form a diazonium compound. This compound is then coupled with N-(1-naphthyl) ethylenediamine dihydrochloride to form a reddish-purple azo dye which is measured using a colorimeter. The procedures of both methods are equivalent to those used in EPA-approved colorimetric methods such as Standard Method 4500-NO₂ B [18th, 19th, and 20th Editions]. Accordingly, SEAL Methods 2-013-1-L and 2-013-1-H may be used in place of other NPDES and NPDWR-approved colorimetric methods for determination of nitrite.

methods. If you have any questions regarding these ATP applications, please contact William We greatly appreciate Seal's interest in the development of environmental monitoring Telliard of SASB (202/566-1061) or Herb Brass of OGWDW/TSC (513/569-7936) at your convenience.

Sincerely,

William A. Telliard

Engineering and Analysis Division (4303T) Director of Analytical Methods

The bons Herb Brass, Ph.D.

Office of Ground Water and Drinking Water Technical Support Center (MS-140)

USEPA Regional Administrators (all Regions)

Quality Assurance Managers (all Regions)

ATP Coordinators (all Regions)

Water Management Division Directors (all Regions)

Gregory J. Carroll, USEPA, OGWDW

Lillian Holmes, USEPA, OGWDW/TSC

Robin K. Oshiro, USEPA, EAD

James Boiani, DynCorp, SCC

STAL/Analydical



Web: www.seal-analytical.com

NITRATE-N + NITRITE-N IN DRINKING AND SURFACE WATERS, AND DOMESTIC AND INDUSTRIAL WASTES

AQ2 METHOD NO: EPA-127-A Rev. 5

TYPE OF SAMPLE

This method is applicable to sewage, effluents, raw and drinking waters. This method is approved for the Clean Water Act for use in wastewater compliance monitoring under the National Pollutant Discharge Elimination System (NPDES) and for use in NPDWR (National Primary Drinking Water Regulations) compliance monitoring. The laboratory should consult the most recent Code of Federal Regulations for applicability and approval by the Clean Water and Safe Drinking Water Acts (40 CFR parts 136 and 141, respectively).

SUBSTANCE DETERMINED

Nitrate and nitrite ions

RANGE OF APPLICATION

0.012 - 2.0 mg N/L

LIMIT OF DETECTION BY USEPA PROCEDURE

MDL = 0.003 mg N/L

METHOD PRINCIPLE - USEPA METHOD 353.2

Nitrate is reduced by copperized cadmium to nitrite, which reacts with sulfanilamide to form a diazonium compound which, in dilute phosphoric acid, couples with N-(1-naphthyl)-ethylenediamine dihydrochloride to form a reddish-purple azo dye. This is measured spectrophotometrically at 520 nm. Separate rather than combined nitrate/nitrite values are obtained by running the samples on sequential tests, i.e., first with, and then without the copperized cadmium reduction step.

INTERFERENCES

Since nitrate-nitrogen is dissolved, the sample may be pre-filtered. Low results may be obtained for samples that contain high concentrations of iron, copper or other metals. EDTA is added to the reagents to eliminate this interference. Samples that contain oil and grease may coat the copperized surface of the cadmium. This is eliminated by pre-extracting the sample with an organic solvent. If fouled, the cadmium is regenerated using an automated utility in AQ2 software. Residual chlorine can produce negative interference by decreasing reduction efficiency. Before analysis, samples should be checked and if required, dechlorinated with sodium thiosulfate.

REFERENCES

Oxidized Nitrogen in Waters 1981, HMSO. (Series Methods for the Examination of Waters and Associated Materials)

Methods for the Determination of Inorganic Substances in Environmental Samples, USEPA 600/R-93/100, August 1993: Method 353.2, Revision 2.0.

SEAL Applications Revision Date: March 3, 2006

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UNITED STATES ENVIRONMENTAL PROTECTION AGENCY WASHINGTON, D.C. 20460

OFFIGE OF WATER

DEC 17 2004

Jessica Ammerman Technical Support Director SEAL Analytical, Inc. 1492 W. Mequon Road Mequon, WI 53092

Dear Ms. Ammerman:

The Statistics and Analytical Support Branch (SASB) and the Office of Ground Water and Drinking Water's Technical Support Center (OGWDW/TSC) have completed their technical review of the materials submitted to support approval of SEAL Method EPA-127-A [Revision Date, September 2004]. This method was submitted as an alternate test procedure (ATP) to determine combined nitrate/nitrite or nitrate and nitrite singly for use in wastewater compliance monitoring under National Pollutant Discharge Elimination System (NPDES) regulations and in drinking water compliance monitoring under national primary drinking water regulations (NPDWR).

We are pleased to inform you that, in the judgement of our technical staff, SEAL Method EPA-127-A is an acceptable version of approved revisions of EPA Method 353.2. EPA Method 353.2 is listed at Title 40 of the Code of Federal Regulations (40 CFR) Part 136 for NPDES compliance monitoring* and 40 CFR Part 141 for NPDWR compliance monitoring. Seal Method EPA-127-A determines combined nitrate/nitrite or nitrate and nitrite singly by colorimetry using discreet analysis. A filtered sample is passed through a copperized-cadmium column to reduce nitrate to nitrite. The nitrite originally present in the sample plus reduced nitrate reacts with sulfanilamide under acidic conditions to form a diazonium compound. This compound is then coupled with N-(1-naphthyl) ethylenediamine dihydrochloride to form a reddish-purple azo dye that is measured using a colorimeter. Separate, rather than combined nitrate/nitrite results, are obtained by carrying out the procedure first with, and then without, the copper-cadmium reduction step (to measure nitrate/nitrite and nitrite respectively)and subtracting the result obtained for nitrite from the result obtained for combined nitrate/nitrite to obtain the result for nitrate. The procedures used are equivalent to those used in EPA-approved colorimetric methods such as EPA Method 353.2. Accordingly, SEAL Method EPA-127-A [Revision Date, September 2004] may be used in place of other NPDES and NPDWR-approved colorimetric methods for determination of combined nitrate/nitrite or nitrate and nitrite singly.

^{*} EPA Method 353.2 is not currently listed at 40 CFR 136 for determination of nitrate and nitrite singly. However,

We greatly appreciate SEAL's interest in the development of environmental monitoring methods. If you have any questions regarding this ATP application (Case No.'s D04-007 and N04-007), please contact William Telliard of SASB (202/566-1061) for wastewater or Herb Brass of OGWDW/TSC (513/569-7936) for drinking water at your convenience.

Sincerely,

William A. Telliard

Director of Analytical Methods

Engineering and Analysis Division (4303T)

Herb Brass, Ph.D.

Technical Support Center (MS-140)

Office of Ground Water and Drinking Water

cc:

USEPA Regional Administrators (all Regions)

Quality Assurance Managers (all Regions)

ATP Coordinators (all Regions)

Water Management Division Directors (all Regions)

Gregory J. Carroll, USEPA, OGWDW

Lillian Holmes, USEPA, OGWDW/TSC

Robin K. Oshiro, USEPA, EAD

Danielle Carter, CSC, SCC

enclosure:

Seal Method: Determination of Nitrate/Nitrite in Drinking and Surface Waters, and Domestic and Industrial Wastes by Discrete Automated Colorimetry Using the Seal AQ2 Analyzer, September 2004.





Web: www.seal-analytical.com

Method Comparison Table for Nitrate + Nitrite (cadmium reduction), Rev. 2

	NITRATE + NITRITE (AA II) AQ2 Nitrate + Nitrite (1 June 2005)				
i	EPA Method 353.2 (March 1983)	AQ2 Nitrate + Nitrite (1 June 2005)			
Scope and Application	Determination of nitrite singly or nitrite and nitrate combined in drinking, surface and saline waters, domestic and industrial wastes. The applicable range is 0.05 to 10.0 mg N/L.	Method No. EPA-126-A Rev 4 Same. The applicable range is 0.03 to 4.5 mg N/L.			
Summary of Method	Nitrate is reduced by copperized cadmium to nitrite which reacts with sulfanilamide to form a diazonium compound which, in dilute phosphoric acid, couples with N-(1-naphthyl) ethylenediamine dihydrochloride to form a reddish-purple azo dye. This is measured spectrophotometrically at 520 nm. Separate, rather than combined nitrate-nitrite values can be obtained by carrying out the procedure both with and without the copperized-cadmium reduction step.	Same			
Interferences (Sample Pre-treatment)	Sample may be pre-filtered to remove particulates. Low results may be obtained for samples that contain high concentrations of iron, copper or other metals. EDTA is added to the reagents to eliminate this interference. Samples that contain oil and grease may coat the copperized surface of the cadmium. This is eliminated by pre-extracting the sample with an organic solvent. If fouled, the cadmium is regenerated using an automated utility in AQ2 software. Residual chlorine can produce negative interference by limiting reduction efficiency. Before analysis, samples should be checked and, if required, dechlorinated with sodium thiosulfate.	Same.			
Reaction Conditions	Residence time in cadmium column is unspecified. Colorimetric residence time in open tubular reactor coil, room temperature, is unspecified.	Cadmium reduction time is 25 sec. Colorimetric reaction time is 420 sec, in static reaction well heated to 37°C			

SEAL Applications Revision Date: August 1, 2005

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	NITRATE + NITRITE (AA II)	AO2 Nika-ka a Nika ini			
	EPA Method 353.2 (March 1983)	AQ2 Nitrate + Nitrite (1 June 2005)			
Equipment	Automated continuous flow analysis	Method No. EPA-126-A Rev 4			
	equipment designed to deliver and react	SEAL AQ2 automated discrete analyzer Configured to deliver and larger			
	sample and reagents in the required order	configured to deliver and react sample and reagents in equivalent order and ratios as			
	and ratios.	USEPA method 353.2.			
	 Copperized-cadmium reduction column, 	Open tubular copperized-cadmium reductor			
	specific to the flow analyzer being used.	coil for use on the SEAL AQ2.			
Reagents and Standards	Ammonia chloride EDTA Buffer: 85 g	Ammonia chloride EDTA Buffer: 134 g			
Statituarus	NH ₄ Cl and 0.1 g Na ₂ EDTA per liter. Adjust	NH₄Cl and 0.3 g Na₂EDTA per liter. Adjust			
	pH to 8.5 using concentrated NH ₄ OH. 0.015% (v/v) Brij-35 [®] surfactant to stabilize	pH to 8.5 using concentrated NH ₂ OH, Add			
	flow through the cadmium column.	0.02% (w/v) Triton X-100® to stabilize flow			
	in an angle the seaming column,	through the cadmium reductor coil.			
	Sulfanilamide-NEDD Color Reagent:	Sulfanilamide-NEDD Color Reagent:			
	100 mL concentrated H ₃ PO ₄ , 40 g	38 mL concentrated H ₃ PO ₄ , 15 g			
	sulfanilamide, 2 g N-(1-naphthyl)	sulfanilamide, 0.75 g N-(1-naphthyl)			
	ethylenediamine dihydrochloride per liter.	ethylenediamine dihydrochloride per liter.			
	Standards (mg N/L):	Consideration (van \$10)			
	0.0, 0.05, 0.1, 0.2, 0.5, 1.0, 2.0, 4.0, 6.0	Standards: (mg N/L)			
	1	0, 0.03, 0.06, 0.135, 0.36, 0.45, 0.9, 2.25, 3.6, 4.5			
Method	Precision - Inter-lab study: mg N/L	Precision, Single Lab: mg N/L			
Performance	Mean %Recovery Std Dev % RSD	11/- //			
	0.29 5.8% 0.012 4.1%	0.073 93.3 0.0040 5.9%			
	0.35 118.1% 0.092 26.3%	0.090 104.4 0.0033 1.5%			
	2.31 104.5% 0.318 14.8%	0.90 105.9 0.013 1.4%			
	2.48 97.3% 0.176 7.1%	3.65 96.5 0.082 2.3%			
	Single-Lab precision is estimated at 50% to				
	75% of Inter-Laboratory precision.				
:	··				
	Method Detection Limit:	Method Detection Limit:			
Final Ratios	No data	0.006 mg N/L			
of Reagents	Reag. Conc. x Mixing Ratio = Rxn.	Reag. Conc. x Mixing Ratio = Rxn.			
3	Conc. Cadmium reduction (pH 8.5)	Cadmium reduction (pH 8.5)			
	NH_4CI 85 g/L x 0.789 = 67 g/L	NH ₄ Cl 134 g/L x 0.50 = 67 g/L			
	· ·	10 1 g/L × 0.00 = 07 g/L			
	Final colorimetric reaction	Final colorimetric reaction			
	NH_4CI 67 g/L x 0.742 = 49.7 g/L	NH_4CI 67 g/L x 0.30 = 20.2 g/L			
		plus (134 g/L x 0.312) x 0.70 = $\underline{29.3 \text{ g/L}}$			
		49.5 g/L			
	H_3PO_4 100 mL/L x 0.258 = 25.8 mL/L	H_3PO_4 38 mL/L x 0.70 = 26.6 mL/L			
	Sulfanil- $40 \text{ g/L} \times 0.258 = 10.3 \text{ g/L}$	Sulfanil- $15 \text{ g/L} \times 0.70 = 26.6 \text{ mL/L}$			
	amide	amide			
	NEDD 2 g/L x $0.258 = 0.52$ g/L	NEDD $0.75 \text{ g/L} \times 0.70 = 0.53 \text{ g/L}$			

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NITRITE-N IN DRINKING AND SURFACE WATERS, AND DOMESTIC AND INDUSTRIAL WASTES

AQ2 METHOD NO: EPA-115-A Rev. 3

TYPE OF SAMPLE

This automated method is applicable to drinking, surface and saline waters, and industrial and domestic wastes. This method is approved for the Clean Water Act for use in wastewater compliance monitoring under the National Pollutant Discharge Elimination System (NPDES) and for use in NPDWR (National Primary Drinking Water Regulations) compliance monitoring. The laboratory should consult the most recent Code of Federal Regulations for applicability and approval by the Clean Water and Safe Drinking Water Acts (U.S. EPA, 40 CFR parts 136 and 141, respectively).

SUBSTANCE DETERMINED

Nitrite ion

RANGE OF APPLICATION

0.015 - 1.2 mg N/L

METHOD DETECTION LIMIT BY USEPA PROCEDURE

MDL = 0.0025 mg N/L

METHOD PRINCIPLE - USEPA METHOD 353.2

Nitrite ions react with sulfanilamide to form a diazonium compound which, in dilute phosphoric acid, couples with N-(1-naphthyl)-ethylenediamine dihydrochloride to form a reddish-purple azo dye. The absorbance of this complex is measured spectrophotometrically at 520 nm.

INTERFERENCES

There are few significant interferences at concentrations less than 1000 times that of the nitrite. Presence of strong oxidants or reductants in the samples may affect the nitrite determination. Highly alkaline samples may give low results due to the shift in pH. For a full discussion of interferences, see the references below.

REFERENCES

Oxidized Nitrogen in Waters 1981, HMSO. (In the series Methods for the Examination of Waters and Associated Materials)

Methods for Chemical Analysis of Waters and Wastes, USEPA, 1983: Method 353.2

Methods for the Determination of Inorganic Substances in Environmental Samples, USEPA 600/R 93/100, 1993: Method 353.2. Revision 2.0.

SEAL Applications Revision Date: June 1, 2005

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Method Comparison Table for NITRITE, Rev 2

	NITRITE (AA II) AQ2 NITRITE (1 June 2005)				
		AQ2 NITRITE (1 June 2005)			
Scope and	EPA Method 353.2 (March 1983)	Method No. EPA-115-A Rev 3			
Application	Determination of nitrite in surface and saline	Same. The applicable range is 0.015 to			
Application	waters, domestic and industrial wastes. The	1.2 mg N/L.			
Summary of	applicable range is 0.05 – 10 mg N/L.				
Method	Nitrite reacts with sulfanilamide to form a	Same			
INCUIOG	diazonium compound which, in dilute				
	phosphoric acid, couples with N-(1-naphthyl)				
	ethylenediamine dihydrochloride to form a reddish-purple azo dye. The absorbance is				
	measured at 520 nm.				
Interferences	Strong oxidants or reductants may degrade				
	nitrite. Highly alkaline samples may impede	Same			
	color formation.				
Equipment	Automated continuous flow analysis	SEAL AGO Sutomata to the season of the seaso			
1	equipment designed to deliver and react	SEAL AQ2 automated discrete analyzer			
	sample and reagents in the required order	configured to deliver and react sample and			
	and ratios.	reagents in equivalent order and ratios as USEPA method 353.2.			
Reagent	Ammonium Chloride Buffer:	Ammonium Chloride Buffer:			
Preparation	85 g NH₃Cl and 0.1 g Na2EDTA per Liter.	134 g NH₃Cl and 0.3 g Na2EDTA per Liter.			
	Adjust to pH 8.5 concentrated NH ₄ OH.	Adjust to pH 8.5 concentrated NH ₄ OH.			
		7 P			
	Sulfanilamide-NEDD Color Reagent:	Sulfanilamide-NEDD Color Reagent:			
	100 mL concentrated H ₃ PO ₄ , 40 g	38 mL concentrated H ₃ PO ₄ , 15 g			
	sulfanilamide, 2 g N-(1-naphthyl)	sulfanilamide, 0.75 g N-(1-naphthyl)			
	ethylenediamine dihydrochloride per liter.	ethylenediamine dihydrochloride per liter			
		pH 8.5 buffer added at 31.2% (v/v).			
	Standards (mg N/L):				
	0.0, 0.05, 0.1, 0.2, 0.5, 1.0, 2.0, 4.0, 6.0	Standards (mg N/L):			
Final Ratios of	Reag. Conc. x Mixing Ratio = Rxn.	0, 0.015, 0.03, 0.06, 0.15, 0.3, 0.6, 0.96, 1.2			
Reagents	Conc.	Reag. Conc. x Mixing Ratio = Rxn.			
•	Ammonium Chloride Buffer:	Ammonium Chloride Buffer:			
	NH_4CI (85 g/L x 0.789) x 0.742 = 49.7 g/L				
	, , , , , , , , , , , , , , , , , , , ,	NH_4Cl 134 g/L x 0.134 = 18.0 g/L plus (134 g/L x 0.312) x 0.671 = 28.0 g/L			
		46.0 g/L			
		46.0 g/L			
	Color Reagent:				
	H_3PO_4 100 mL/L x 0.258 = 25.8 mL/L	H_3PO_4 38 mL/L x 0.671 = 25.5 mL/L			
	Sulfanil- 40 g/L x 0.258 = 10.3 g/L amide	Sulfanil- $15 \text{g/L} \times 0.671 = 10.5 \text{g/L}$			
	A 1 Programs	amide			
	NEDD 2 g/L x $0.258 = 0.52$ g/L	NEDD $0.75 \text{ g/L} \times 0.671 = 0.50 \text{ g/L}$			

SEAL Applications Revision Date: August 1, 2005

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	NITRITE (AA II) EPA Method 353.2 (March 1983)	AQ2 NITRITE (1 June 2005) Method No. EPA-115-A Rev 3 Static reaction well heated to 37°C, with incubation time 10 min.			
Reaction Conditions	Flow-through coiled reactor, room temperature, with unspecified residence time.				
Method Performance	Precision and accuracy data: Not available	Precisio Known 0.025 0.60	%Recovery 96.0 103.1	(mg NO ₂ -N/L Std. Dev. 0.0006 0.0051	%RSD 2.5% 0.8%
	Method Detection Limit: No data	Method Detection Limit: 0.0025 mg NO ₂ -N/L			

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NITRITE-N IN DRINKING AND SURFACE WATERS, AND DOMESTIC AND INDUSTRIAL WASTES

AQ2 METHOD NO: EPA-137-A Rev. 1

TYPE OF SAMPLE

This low-range method is applicable to drinking, surface and saline waters, and industrial and domestic wastes. This method is approved for the Clean Water Act for use in wastewater compliance monitoring under the National Pollutant Discharge Elimination System (NPDES) and for use in NPDWR (National Primary Drinking Water Regulations) compliance monitoring. The lab should consult the most recent Code of Federal Regulations for applicability and approval by the Clean Water and Safe Drinking Water Acts (40 CFR parts 136 and 141, respectively).

SUBSTANCE DETERMINED

Nitrite ion

RANGE OF APPLICATION

0.002 - 0.2 mg N/L

METHOD DETECTION LIMIT BY USEPA PROCEDURE

MDL = 0.0006 mg N/L

METHOD PRINCIPLE – USEPA METHOD 353.2

Nitrite ions react with sulfanilamide to form a diazonium compound which, in a phosphoric acid/dihydrogen phosphate buffer, couples with N-(1-naphthyl)-ethylenediamine dihydrochloride to form a reddish-purple azo dye. The absorbance of this complex is measured at 520 nm. To achieve lower detection limits than USEPA method 353.2, this method employs 6-fold reduced quantity of color reagent. However, the same relative ratios among reactants are maintained

INTERFERENCES

There are few significant interferences at concentrations less than 1000 times that of the nitrite. Presence of strong oxidants or reductants in the samples may affect the nitrite determination. Highly alkaline samples may give low results due to the shift in pH. The phosphate buffer will compensate for total alkalinity at least 1500 mg CaCO₃/L. For a full discussion of interferences, see the references below.

REFERENCES

Oxidized Nitrogen in Waters 1981, HMSO.

(In the series Methods for the Examination of Waters and Associated Materials)
Methods for Chemical Analysis of Waters and Wastes, USEPA, 1983: Method 353.2
Methods for the Determination of Inorganic Substances in Environmental Samples, USEPA 600/R 93/100, 1993: Method 353.2, Revision 2.0.

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UNITED STATES ENVIRONMENTAL PROTECTION AGENCY

CINCINNATI, OHIO 45268

Jessica Ammerman jammerman@seal-us.com Technical Support Director SEAL Analytical 1492 Mequon Road Mequon, WI 53092

03/14/2008

RE: ATP Case Nos. D06-0033 and D06-0034

Dear Ms. Ammerman:

Per the terms of the Alternate Test Procedure (ATP) program, the Office Ground Water and Drinking Water's Technical Support (OGWDW/TSC) has determined that SEAL Analytical AQ2 Method EPA-126-A [Revision 5] for the determination of nitrate and nitrite and SEAL AQ2 Method EPA-137-A [Revision 1] for the determination of nitrite are acceptable versions of EPA Method 353.2 [Revision 2.0]. Accordingly, SEAL AQ2 Method EPA-126-A [Revision 5] and SEAL AQ2 Method EPA-137-A [Revision 1] may be used for compliance monitoring performed under National Primary Drinking Water Regulations. SEAL AQ2 Method EPA-126-A [Revision 5] and SEAL AQ2 Method EPA-137-A [Revision 1] employ discreet analysis and rely on the same chemistry and determinative technique as those specified in EPA Method 353.2 [Revision 2.0] to determine nitrate and nitrite.

We appreciate your interest in the development of environmental monitoring methods. If you have any questions regarding the review of these alternate test procedures (ATP Case Nos. D06-0033 and D06-0034), please contact Steve Wendelken by e-mail at: wendelken.steve@epa.gov or by telephone at: 513-569-7491.

Sincerely,

Steven C. Wendelken, Ph.D.

ATP Coordinator

Office of Ground Water and Drinking Water

Technical Support Center (MS-140)

26 W. Martin Luther King Dr.

Cincinnati, Ohio 45219

Phone: (513) 569-7491

Fax: (513) 569-7837

wendelken.steve@epa.gov

cc:

ATP Coordinators (all Regions)
Quality Assurance Managers (all Regions)
Gregory J. Carroll, USEPA, OGWDW
Danielle Carter, CSC, SCC





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Method Comparison Table for NITRITE, Rev 1

	NITRITE (AA II)	A O O A SET POLITICAL OF THE A O O A SET POLITICAL OF THE A O O A SET POLITICAL OF THE A O O O O O O O O O O O
	, ,	AQ2 NITRITE (13 February 2006)
6	EPA Method 353.2 (March 1983)	Method No. EPA-137-A Rev 1
Scope and	Determination of nitrite in surface and saline	Same. The applicable range is 0.002 to
Application	waters, domestic and industrial wastes. The	0.2 mg N/L.
Summary of	applicable range is 0.05 – 10 mg N/L.	
Method	Nitrite reacts with sulfanilamide to form a	Same
Meanoa	diazonium compound which, in dilute	
	phosphoric acid/ dihydrogen phosphate buffer, couples with N-(1-naphthyl)	
	ethylenediamine dihydrochloride to form a	
	reddish-purple azo dye. The absorbance is	
	measured at 520 nm.	
Interferences	Strong oxidants or reductants may degrade	Same. The goldie phoenhate buffer in
	nitrite. Highly alkaline samples may impede	Same. The acidic phosphate buffer in the reagent system (designed per EPA 353.2) will
	color formation.	compensate for total alkalinity at least
		1500 mg CaCO ₃ /L.
Equipment	Automated continuous flow analysis	SEAL AQ2 automated discrete analyzer
	equipment designed to deliver and react	configured to deliver and react sample and
	sample and reagents in the required order	reagents in equivalent order and ratios as
	and ratios.	USEPA method 353.2.
Reagent	Ammonium Chloride Buffer:	Sulfanilamide-NEDD Reagent (combined):
Preparation	85 g NH₄Cl and 0.1 g Na₂EDTA per Liter.	4 g NaOH, 2 g Na ₂ EDTA, 100 mL
	Adjust to pH 8.5 concentrated NH ₄ OH.	concentrated H₃PO₄, 100 g NH₄Cl, 25 g
	Sulfanilamida NEDD Calar Danner	sulfanilamide, 1.25 g N-(1-naphthyl)
	Sulfanilamide-NEDD Color Reagent: 100 mL concentrated H ₃ PO ₄ , 40 g	ethylenediamine dihydrochloride per liter.
	sulfanilamide, 2 g N-(1-naphthyl)	The 12 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1
	ethylenediamine dihydrochloride per liter.	To achieve lower detection limits than USEPA
	any aroundide per mer,	method 353.2, this method employs 6-fold
		reduced quantity of color reagent.
	Standards (mg N/L):	Standards (mg N/L):
	0.0, 0.05, 0.1, 0.2, 0.5, 1.0, 2.0, 4.0, 6.0, 10.0	0, 0.002, 0.006, 0.016, 0.050, 0.1, 0.15, 0.2
Reaction	Flow-through coiled reactor, room	Static reaction well heated to 37°C, with
Conditions	temperature, with unspecified residence time.	incubation time 10 min.

SEAL Applications Revision Date: February 13, 2006

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	AUTPITE (AA II)				
	NITRITE (AA II)	AQ2 NITRITE (13 February 2006)			
	EPA Method 353.2 (March 1983)	Method No. EPA-137-A Rev 1			
Final Reagent	Reag. Conc. x Mixing Ratio = Rxn.	Reag. Conc. x Mixing Ratio = Rxn.			
Concentration	Conc.	Conc.			
	Ammonium Chloride Buffer:	Color Reagent:			
	NH ₄ CI (85 g/L x 0.789) x 0.742 = 49.7 g/L	NH_4CI 100 g/L × 0.0654 = 6.54 g/L			
	Provides pH buffering in final reaction mixture	NaOH $4 g/L \times 0.0654 = 0.262 g/L$			
	Color Reagent:	Addition of NaOH provides pH buffering.			
	Joint Magaire.				
	H_3PO_4 100 mL/L x 0.258 = 25.8 mL/L	H ₃ PO ₄ 100 mL/L x 0.0654 = 6.54 mL/L			
	Sulfanil- 40 g/L x 0.258 = 10.3 g/L	Sulfanil- 25 g/L x 0.0654 = 0.54 mL/L Sulfanil- 25 g/L x 0.0654 = 1.63 g/L			
	amide	amide 20 9/2 x 0.0004 = 1.03 9/2			
	NEDD 2 g/L x 0.258 = 0.52 g/L	NEDD 1.25 g/L x 0.0654 = 0.082 g/L			
Final Ratios	H₃PO₄ / Sulfanilamide / NEDD:	H₃PO₄ / Sulfanilamide / NEDD:			
of Reagents	2.5 mL/L / 1.0 g/L / 0.05 g/L	4.0 mL/L / 1.0 g/L / 0.05 g/L			
	NILL CI / Cultion Hamilton	_			
	NH₄Cl / Sulfanilamide: 4.8 g/L_/ 1.0 g/L	NH ₄ Cl / Sulfanilamide:			
Method	Precision and accuracy data:	4.0 g/L / 1.0 g/L			
Performance	Not available	Precision, Single Lab (mg NO ₂ -N/L			
		Known %Recovery Std. Dev. %RSD			
		blank 0.0003			
		0.002 n.nnos I			
		0.01 118.5 0.0005 3.9%			
		0.2 104.5 0.0025 1.2%			
	Mathed Detection 1 to 1				
	Method Detection Limit: No data	Method Detection Limit:			
	INU Udid	0.0006 mg NO ₂ -N/L			

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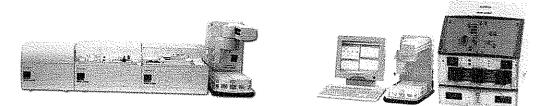
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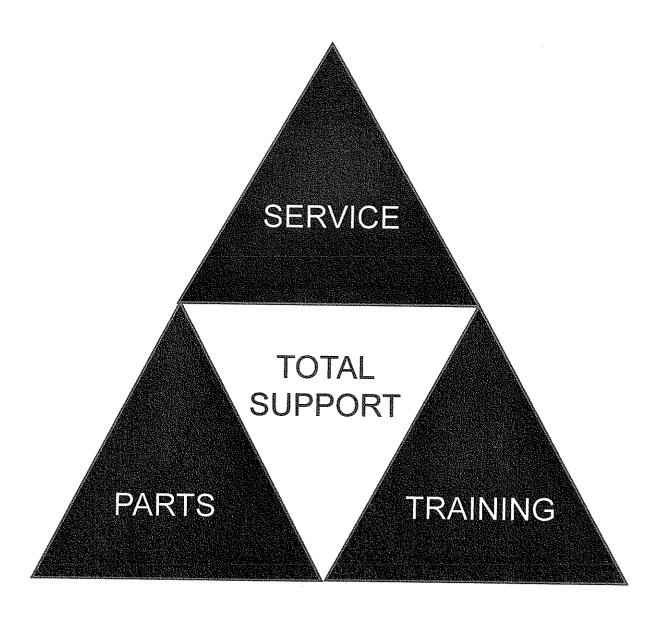
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THE TECHNICAL TEAM

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BASIC AGREEMENT

This is an entry-level agreement which offers you special privileges when you need service during the year. In exchange for your commitment to utilize our on-site service, we will extend to you the following benefits:

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9 0

Base Annual Premium: \$5400

Note 1: Discounts are available for multiple systems on single or multiple sites.

Excludes PC, PC printer and manifolds

Tel: 262-241-7900 Fax: 262-241-7970

Website: www.seal-analytical.com

Fax To: 262-241-7970



CUSTOMIZED AGREEMENT

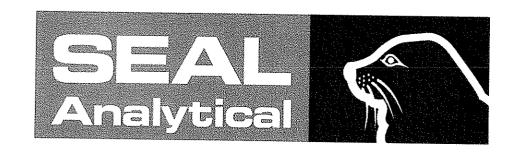
Attn: Lalicia Potter Should the Basic or Premium Agreement not fulfil your needs, SEAL Analytical offers you the opportunity to create your own specific requirements. Just complete the following. Name: Company: Address: Tel:.....Fax:.... Email:.... I require the following to be included in the Service Support Scheme for my..... Please Select Service parts at discount (Please specify %) All service parts included One preventative maintenance visit More than one - state how many One emergency visit More than one - state how many Unlimited telephone and email support Consumables at discount (Please specify %) Operator training at discount (Please specify %) Access to applications library Free software upgrades

Other (Please specify)

SEAL Analytical, Inc.
Mequon Technology Center
10520-C Baehr Rd.
Mequon, WI 53092
Phone: 262 244 7000

Phone: 262-241-7900 Fax: 262-241-7970

Email: sales@seal-us.com www.seal-analytical.com



References for SEAL Analytical, Inc.

1. Gainesville Regional Utilities

3901 SW 63rd Blvd. Gainesville, FL 32608 Phone: (352) 393-6729

Contact: Ms.Sandy Barnes, Senior Laboratory Technician Purchased AQ2 Multi-Chemistry Analyzer – 1/13/05

2. City of Newport News

NNW WQC Lab 3629 G.W. Memorial Hwy. Yorktown, VA 23693 Phone: (757) 234-6723

Contact: Ms. Sherry Williams

Purchased AQ2 Multi-Chemistry Analyzer - 2/28/03

3. Las Vegas Valley Water District

1299 Burkholder Blvd. Henderson, NV 89015 Phone: (702) 856-3520

Contact: Stan Van Wagenen, Lab Director

Purchased AQ2 Multi-Chemistry Analyzer - 05/10/04

4. Microbac Laboratories

3323 Gilmore Industrial Blvd.

Louisville, KY 40213 Phone: (502) 962-6400

Contact: John Miller

Purchased AQ2 Multi-Chemistry Analyzer - 05/26/05

5. Environmental Chemists

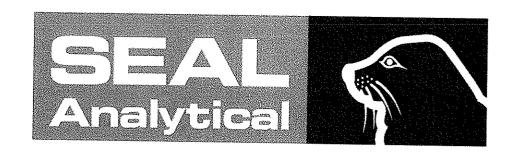
6602 Windmill Way Wilmington, NC 28405 Phone: (910) 392-0223

Contact: Ray Porter

Purchased AQ2 Multi-Chemistry Analyzer - 04/16/08

Phone: 262-241-7900 Fax: 262-241-7970

Email: sales@seal-us.com www.seal-analytical.com



Qualifications and Experience of Technical Staff

Stuart Smith, President

- Educational background in chemistry and electronics
- Over 20 years experience in the industry, including over 10 years at Bran+Luebbe UK and SEAL Analytical UK
- **>** Phone: (262) 241-7900, ext. 214
- **⊃** Email: <u>ssmith@seal-us.com</u>

Tim Bahowick, Applications Director

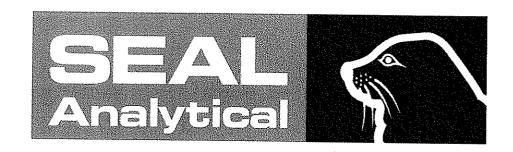
- ⇒ PhD in Chemistry
- Over 13 years experience in the industry, including 6 years at Lachat Instruments and 7 years at SEAL Analytical, Inc.
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- ➡ Email: tbahowick@seal-us.com

Lalicia Potter, Technical Sales & Support Director

- Graduate of the University of Sydney, AU with Honors degree in chemistry
- Over 15 years experience in the industry, including over 10 years at Bran+Luebbe Australia and Bran+Luebbe UK
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- **⊃** Email: <u>lpotter@seal-us.com</u>

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- Masters of Science in Chemistry
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Katie Parkhurst, Technical Support Chemist

Bachelor of Science in Chemistry

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Mark Kresa, Technical Support Service Technician

Graduate of ITT Technical Institute

Over 10 years experience in the industry, including over 5 years at Lachat Instruments and 5 years at Seal Analytical, Inc.

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⊃ Email: <u>mkresa@seal-us.com</u>

Sam Dorsey, Technical Support Service Technician

➡ Graduate of Utah Valley State University in Electronics Technology

 Over 21 years experience in the industry, providing direct repair service and training to Bran+Luebbe customers

> Phone: (262) 241-7900, ext. 206

⊃ Email: sdorsey@seal-us.com

Paul Gundersen, Customer Service Specialist

Over 21 years experience in Inventory management and direct customer service

> Phone: (262) 241-7900, ext. 210

⇒ Email: parts@seal-us.com

NOTE: All are direct employees of SEAL Analytical, Inc.



Jeff TeRonde Controller



Stuart Smith President



Lalicia Potter Technical Sales Director



Corrin Breitigam Technical Support Chemist

Who's Who SEAL USA



Tim Bahowick Ph.D Technical Applications Director



Katie Parkhurst Technical Support Chemist

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Mark Kresa Technical Support Specialist



Sam Dorsey Technical Support Specialist



Paul Gundersen Customer Service Specialist

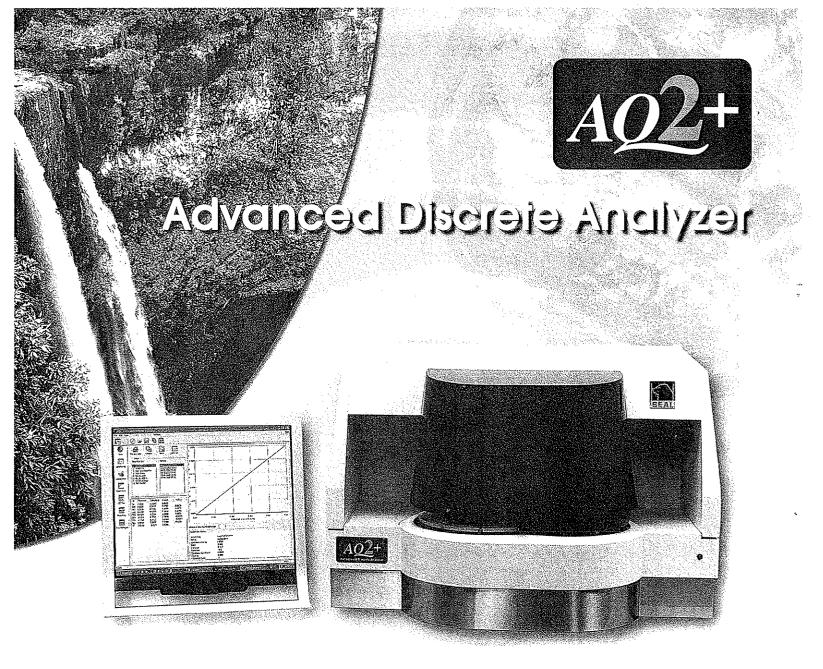


Jeanne Kimble Senior Administrator



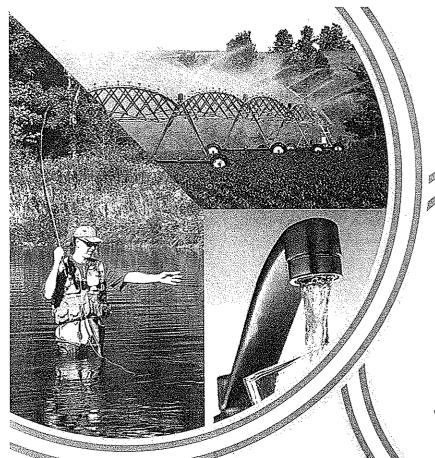
Nancy Vallette Administrator





Highly Automated
Simple to Use
Lowest Reagent Consumption
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AQ2+ Series...

is a flexible analyzer that uses the principle of discrete analysis where each test occurs in a separate or discrete reaction vessel.

AQ2+ is Ideal...

when many and varied tests are needed on different samples and/or individual results are needed in a short time.

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The AQ2+ employs a robust robotic sampling arm working in conjunction with a stepper motor-driven syringe that is responsible for aspirating, dispensing and mixing accurate and precise quantities of sample and reagent in miniaturized test tubes called reaction wells.

The sample and reagents are incubated in the reaction wells for a preprogrammed time. A single aliquot is then transferred into a glass optical cuvette. The absorbance is then read on the stationary reactant to ensure the best possible signal to noise ratio.

The reaction times are user programmable from seconds to minutes. The standard methods supplied have optimized reaction times which ensure the reaction is brought to completion. This complete reaction emulates the manual and segmented flow methods. This eliminates the problems encountered with flow injection analysis where the reaction is not brought to completion, decreasing sensitivity and potentially introducing kinetic effects.

Once the absorbance is read, the cuvette is thoroughly cleaned, eliminating any carry over or cross contamination.



AQ2+ Advantages and Benefits

The AQ2+ Advanced Discrete Analyzer is a fully automated instrument designed specifically for the environmental market.



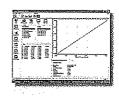
- 100% optical quality glass cuvette used for precise absorbance measurement
- Robust detection system utilizes stationary measurement cell
- 10 mm optimum path length



- Reagent wedges with on-board cooling only 20-400 µL reagent used per test
- Automatic reagent level sensing verifies sufficient reagent volume
- Reagent expiration date tracked through software



- Low cost, disposable wells used for each discrete reaction
- Constant heating and programmable reaction time ensure reaction reaches completion



- Highly flexible software designed with user input
- QCProTM Data Quality Assurance
 System allows the user to specify
 QC types, limits and corrective
 actions upon a QC failure



- Integrated automated cadmium coil reduction for nitrate/nitrite determination
- In situ coil regeneration is fully automated through software



- Both 57 and 100-position sample trays are available to accommodate sample load
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and other international regulatory compliant methods are available. Please visit www.seal-analytical.com or contact your local SEAL Analytical representative about your specific application.

METHODS INCLUDE:

Alkalinity

Phenolics

Ammonia

Phosphate, ortho

Chloride Cvanides Phosphorus, total Silicate

Nitrate/Nitrite

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Nitrite

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Walk away operation
- including ability to run overnight

Segregated chemical waste and wash minimizes environmental impact

Only 500 – 600 µL of combined reagents and sample consumed per test

Add samples after a run has started

Automated standard preparation and dilution of over range samples

Will run up to 150 tests per hour

Tests programmable per sample to reduce analysis time

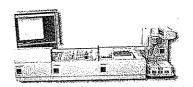


Comprehensive Support

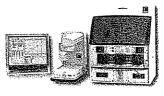
We offer comprehensive applications, technical, service and software support prior to and following installation and training. *These include...*

- Development of custom chemistries
- Adaptation of existing methods to specific requirements such as matrix, range or detection limit adjustments
- Guaranteed availability of genuine consumables and spare parts
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QuAAtro

The AutoAnalyzer 3 HR and QuAAtro are the latest generation segmented flow analyzers. Based on the robust, world class heritage of the original Technicon ™ design, these instruments offer exceptional reliability, performance and ease of use for over 1000 available methods.

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Both instruments are especially suitable for ultra low level determinations, such as drinking water and in complex matrices such as seawater with varying salinity.

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